LITERATURE CITED

- 1. J. Folch, M. Lees, and J. H. Sloane-Stanley, J. Biol. Chem., 226, 497 (1957).
- 2. M. E. McKillican and J. A. G. Larose, J. Amer. Chemists' Soc., 47, No. 7, 256 (1970).
- 3. Methods of Carbohydrate Chemistry [in Russian], Moscow (1967).
- 4. É. V. Dyatovitskaya, T. I. Torkhovskaya, and L. D. Bergel'son, Biokhimiya, 34, 117 (1969).

FLAVONOIDS OF Senecio subdentatus. III

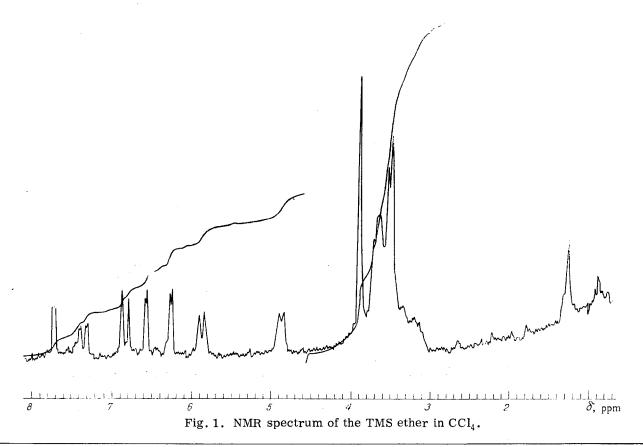
T. K. Chumbalov, O. V. Fadeeva, and T. K. Nikishchenko

UDC 574.972

We have isolated two flavonol glycosides from the herb Senecio subdenatus L. D. B.

Compound 1, $C_{28}H_{32}O_{17}$ mp 210-212°C, R_f 0.31 (BAW, 4:1:2); 0.56 (15% acetic acid), $[\alpha]_D^{20}-85^\circ$ (c 0.5; DMFA), λ_{max} 355, 255 nm (log ϵ 4.15, 4.28). On hydrolysis with 5% sulfuric acid, it was split into isorhamnetin (yield 52%), galactose, and glucose.

Compound 2, $C_{27}H_{30}O_7$, mp, 215-217°C, $[R_f\ 0.29/0.55,\ [\alpha]_D^{20}-92^\circ\ (c\ 0.3;\ DMFA),\ \lambda_{max}\ 360,\ 257\ nm\ (log\ \epsilon\ 4.13,\ 4.31);$ was hydrolyzed to quercetin (yield 49%), galactose and glucose.



S. M. Kirov Kazakh State University, Alma-Ata State Medical Institute. Translated from Khimiya Prirodnykh Soedinenii, No. 6, p. 802, November-December, 1975. Original article submitted February 25, 1975.

This material is protected by copyright registered in the name of Plenum Publishing Corporation, 227 West 17th Street, New York, N.Y. 10011. No part of this publication may be reproduced, stored in a retrieval system, or transmitted, in any form or by any means, electronic, mechanical, photocopying, microfilming, recording or otherwise, without written permission of the publisher. A copy of this article is available from the publisher for \$7.50.

The isorhamnetin and quercetin were identified by their physicochemical constants and by a comparison with markers [1]. The presence of the sugars in the glycosides at C_3 and C_7 was established by a spectral investigation in the UV region.

The products of stepwise acid hydrolysis (0.1% hydrochloric acid) were 7-monosides identical with compounds isolated previously: isorhamnetin 7-O- β -D-glucopyranoside, C₂₂H₂₃O₁₂, mp 250-252°C, [α]²⁰_D-40.5°, and quercetin 7-O- β -D-glucopyranoside, C₂₁H₂₀O₁₂, mp 247-248°C [α]²⁰_D-52° [2]. Alkaline hydrolysis [3] (0.5% KOH) yielded isorhamnetin 3-galactoside, C₂₂H₂₃O₁₂, mp 198-200°C, [α]²⁰_D-120° (c 0.1; methanol) and quercetin 3-galactoside, C₂₁H₂₀O₁₂, mp 237-239°C, [α]²⁰_D-128° (c 0.12; methanol), respectively. The results of a calculation of molecular rotations according to Klyne [4] showed the β -linkage and the furanose form of the galactose, and this was confirmed by the results of differential spectroscopy in the UV region (890, 1030, 1070 cm⁻¹) and the rapid acid hydrolysis [5]. The NMR spectrum of the TMS ether in CCl₄ showed that the carbohydrates are attached by β -linkages at the C₃ and C₇ of the aglycone: d, 5.80 ppm (1H, J = 7 Hz), d 4.80 ppm (1H, J = 7 Hz) [6] (Fig. 1). On the basis of the results obtained, the flavonoids isolated can be characterized as 3,4',5,7-tetrahydroxy-3' methoxyflavone 3-O- β -D-galactofuranoside-7-O- β -D-glucopyranoside (I) and 3,3'4',5,6-pentahydroxyflavone 3-O- β -D-galacofuranoside-7-O- β -D-glucopyranoside.

LITERATURE CITED

- 1. T. K. Chumbalov, O. V. Fadeeva, and T. K. Nikishchenko, Khim. Prirodn. Soedin., 390 (1972).
- 2. T. K. Chumbalov, O. V. Fadeeva, and T. K. Nikishchenko, Khim. Prirodn. Soedin., 89 (1974).
- 3. V. I. Litvinenko and V. A. Makarov, Khim. Prirodn. Soedin., 336 (1969).
- 4. W. Klyne, Biochem. J., 47, xli (1950).
- 5. J. T. Edward, Chem. Ind. (London), 1102 (1955).
- 6. T. J. Mabry, K. R. Markham, and M. B. Thomas, The Systematic Identification of Flavonoids, Springer, New York (1970).

FLAVONOIDS OF THE NEEDLES OF Picea ajanensis

S. Z. Ivanova, S. A. Medvedeva,

UDC 547.972

V. I. Lutskii, N. A. Tyukavkina, and N. D. Zelenikina

From a fraction of the methanolis extract of the needles of <u>Picea ajanenis</u> (Yeddo spruce) [1] by preparative chromatography on polyamide sorbent (with chloroform—methanol as the eluent) we have isolated the total flavonoid components. GLC analysis of the TMS ethers in comparison with the authentic samples showed the presence of naringenin (RT 7.12 min, taken as 1.0), aromadendrin (RRT 1.15), kaempferol (RRT 1.86), spigenin (RRT 1.91), and quercetin (RRT 2.56). For identification by the GLC method, we used additions of authentic samples.

The analysis was performed on a Tsvet-4 chromatograph with a flame-ionization detector. The column, 300×0.3 cm, was filled with 5% of SE-30 on Chromaton N-AW-HMDS, carrier gas helium, column temperature 284° C, evaporator temperature 350° C.

The naringenin and apigenin, which were present in the fraction in sufficient amount, were obtained by column chromatography on silica gel (with chloroform-methanol as the eluent). Naringenin - mp 249-251 °C (CH₃OH), λ_{max} 288 nm (log ϵ 4.20), mol. wt. 272 (mass spectrometrically).

Apigenin - mp 346-348°C (CH₃OH), λ_{max} 268, 336 nm (log ϵ 4.39, 4.42), mol. wt. 270 (mass spectrometrically).

Irkutsk Institute of Organic Chemistry, Siberian Branch of the Academy of Sciences of the USSR. Translated from Khimiya Prirodnykh Soedinenii, No. 6, pp. 802, 804, November-December, 1975. Original article submitted March 31, 1975.

This material is protected by copyright registered in the name of Plenum Publishing Corporation, 227 West 17th Street, New York, N.Y. 10011. No part of this publication may be reproduced, stored in a retrieval system, or transmitted, in any form or by any means, electronic, mechanical, photocopying, microfilming, recording or otherwise, without written permission of the publisher. A copy of this article is available from the publisher for \$7.50.